partial hydrolysis, Smith degradation and alkaline degradation techniques. Structures of cellulose, chitin, starch and glycogen. Industrial applications of cellulose & starch. Photosynthesis and Fermentation.

UNIT II

Peptides & Proteins: Peptide bond formation and synthesis of polypeptides, Amino andcarboxy protecting groups in peptide synthesis, Solid phase peptide synthesis-Merrifield method, Peptide structure determination-Sequence and End group analysis (N-Terminal and C-Terminal), Secondary, Tertiary and Quaternary structure of proteins.

Nucleic acids: Nucleosides and Nucleotides, Chemical synthesis of nucleosides and nucleotides.

Poly nucleotides- Structure and functions of DNA and RNA.

Non steroidal hormones: Study of the Oxytocin, Vasopressin and synthetic analogs, General study of ACTH, Growth hormones, Somotropin and Insulin.

UNIT III

[12 Hours]

Vitamins: Classification and Nomenclature. Source, deficiency diseases and biological functions of Vitamins. Study of Vitamin A1, Vitamin B3, Vitamin B5, Vitamin C and Vitamin K1.

Antibiotics: Introduction, Classification, Chemistry of Pencillin V, Streptomycin, Chloramphenicol and Tetracyclin.

REFERENCES:

- 1. Organic Chemistry-P.Y.Bruice (Pearson Education Pvt. Ltd., New Delhi), 2002.
- 2. Organic Chemistry 4th Edn.–S.H. Pine et al (McGraw-Hill, London) 1987.
- 3. Advanced Organic Chemistry- R.A. Carey and R.J. Sundberg (Plenum, New York)1990.
- 4. Organic Chemistry, Vol I & II, I.L.Finar (Longmann ELBS, London), 1973.
- 5. Natural Products Chemistry, Vol-I & II- G.R.Chatwal (Himalaya), 1990.
- 6. Chemistry of Natural Products: A Unified Approach-N R Krishnaswamy (University Press)
- 7. 1999.

Chemistry of Natural Products-Sujata V. Bhat, B.A. Nagasampagi, Meenakshi Sivakumar (Springer-Narosa) 2005.

CH E 506 : ANALYTICAL & GREEN CHEMISTRY

COURSE OUTCOME:

Enable the students:

- To understand the basic principles and theory of UV-Visible, Electronic, Infra Red, Nuclear Magnetic Resonance and Mass Spectroscopy.
- To study the utility of these techniques in structure elucidation of simple organic molecules.
- To know about water cycle, water sources, water quality, significant measurements of water parameters and treatment of water for drinking and industrial purposes.
- To learn about principles and use of green chemistry in laboratory synthesis.
- To understand the basic principles and utility of sonochemistry and Microwave induced organic synthesis.

[12 Hours]

UNIT-I:

UV/Electronic Spectroscopy: Basic principles, Beer-Lambert law, types of absorption bands, Factors affecting the positions of UV bands. Theoretical prediction of \Box max for polyenes, \Box , \Box -unsaturated aldehydes, ketones (Woodward-Fieser rules) and substituted benzenes.

IR Spectroscopy: Basic principles, Application of infrared spectroscopy in the structural study-identity by finger printing and identification of functional groups. Characteristic vibrational frequencies of alkanes, alkenes, alkynes, aromatic compounds, alcohols, ethers, phenols and amines). Study of vibrational frequencies of carbonyl compounds (ketones, aldehydes, esters, amides and acids). Factors affecting band positions and intensities

Nuclear Magnetic Resonance Spectroscopy: Basic principles, Solvents used, chemical shift and its measurements, factors affecting chemical shift. Integration of NMR signals, spin-spin coupling, coupling constant. Shielding and deshielding. High resolution ¹H NMR. Applications of NMR spectroscopy in structure elucidation of simple organic molecules.

Mass Spectrometry: Basic principles, molecular ions, meta-stable ions and isotope ions. Fragmentation processes, McLafferty rearrangement. retro Diels-Alder fragmentations. Nitrogen rule.

UNIT-II:

Hydrologic cycle, sources, chemistry of sea water, criteria and standards of water qualitysafe drinking water, maximum contamination levels of inorganic and organic chemicals, radiological contaminants, turbidity, microbial contaminants. Public health significance and measurement of colour, turbidity, total solids, acidity, alkalinity, hardness, chloride, residual chlorine, sulphate, fluoride, phosphate and different forms of nitrogen in natural and polluted water. Chemical sources of taste and odour, treatment for their removal, sampling and monitoring techniques. Determination and significance of DO, BOD ,COD and TOC. Water purification for drinking and industrial purposes, disinfection techniques, demineralization, desalination processes and reverse osmosis . Treatment of liquid radioactive wastes

UNIT-III:

Green Chemistry: Definition and principles, planning a green synthesis in a chemical laboratory, Green preparation-Aqueous phase reactions, solid state (solventless) reactions, photochemical reactions, Phase transfer catalyst catalysed reactions (Quaternary ammonium salts & Crown ethers), enzymatic transformations & reactions in ionic liquids.

Sonochemistry: Introduction, instrumentation, the phenomenon of cavitation, Sonochemical esterification, substitution, addition, oxidation, reduction and coupling reactions.

Microwave induced organic synthesis: Introduction, reaction vessel and reaction medium, concept, specific effect, atom efficiency, % atom utilisation, advantages and limitations, alkylation of active methylene compounds, N-alkylation, condensation of active methylene compounds with aldehydes, Diels-Alder reaction, Leuckardt reductive amination of ketones, ortho ester Claisen rearrangement.

REFERENCES:

- 1. Organic Spectroscopy-3rd Ed.-W.Kemp (Pagrave Publishers, New York), 1991.
- 2. Spectrometric Identification of Organic Compounds Silverstein, Bassler & Monnill (Wiley)1981.
- 3. Applications of Absorption Spectroscopy of Organic Compounds-Dyer(Prentice Hall,NY) 1965.
- 4. Spectroscopy of Organic Compounds-3rd Ed.-P.S.Kalsi (New Age, New Delhi) 2000.
- 5. Spectroscopic Methods in Organic Chemistry Williams and Fleming, TMH.
- 6. A.K. De : Environmental Chemistry, (Wiley Eastern).
- 7. S.K.Banerji : Environmental Chemistry, (Prentice Hall India), 1993.
- 8. 8 S.D. Faust and O.M. Aly : Chemistry of Water Treatment, (Butterworths), 1983.

[12 Hours]

[12 Hours]